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# Optimization of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> thin films deposited by co-evaporation on polyimide for thermoelectric applications

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## ABSTRACT

The optimization of the deposition process of n-type Bismuth Telluride and p-type Antimony Telluride thin films for thermoelectric applications is reported. The films were deposited on a 25  $\mu$ m-thick flexible polyimide (kapton) substrate by co-evaporation of Bi and Te, for the n-type element, and Sb and Te, for the p-type element. The evaporation rate of each material was monitorized by an oscillating crystal sensor and the power supplied to each evaporation boat was controlled with a PID algorithm in order to achieve a precise user-defined constant evaporation rate.

The influence of substrate temperature (in the range 240–300 °C) and evaporation rates of Bi, Te and Sb on the electronic properties of the films was studied and optimized to obtain the highest Seebeck coefficient. The best n-type Bi<sub>2</sub>Te<sub>3</sub> films were deposited at 300 °C with a polycrystalline structure, a composition close to stoichiometry, electrical resistivity ~20  $\mu\Omega$  m and Seebeck coefficient –195  $\mu$ V/°C. The best p-type Sb<sub>2</sub>Te<sub>3</sub> films were deposited at 240 °C, are slightly Te-rich, have electrical resistivity ~20  $\mu\Omega$  m and Seebeck coefficient +153  $\mu$ V/°C. These high Seebeck coefficients and low electrical resistivities make these materials suitable for fabrication of Peltier coolers and thermopile devices.

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# 1. Introduction

Efficient solid state thermoelectric microdevices are desirable for several applications, including local cooling and thermoelectric microgeneration. Thermoelectric cooling and generation based on Peltier and Seebeck effects, respectively, have the advantages of not using any moving mechanical parts, being environmental friendly, allowing integration with microelectronic circuits and being easy to control. Tellurium alloys (n-type Bi<sub>2</sub>Te<sub>3</sub> and p-type Sb<sub>2</sub>Te<sub>3</sub>) are wellestablished low-temperature thermoelectric materials widely used in the thermoelectric industry due to their high Seebeck coefficient, low electrical resistivity and low thermal conductivity [1].

Because of the reduced dimensions of the microdevices, the conventional processes used to produce bulk thermoelectric materials are not applicable to the present case. Instead, thin film planar technology was considered adequate for this task, due to the large choice of substrates that it allows, to the possibility of patterning the devices to micro or submicrodimensions by the use of masks or photolithography and to the ease of integration with standard Si technology.

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Tentative deposition of Bi<sub>2</sub>Te<sub>3</sub> films by direct evaporation of the bulk materials proved to be impossible due to the large differences in vapour pressure of Bismuth and Tellurium, resulting in a compositional gradient along the film thickness [2]. Other deposition processes have been reported for the deposition of Bi<sub>2</sub>Te<sub>3</sub> thin films. Thermal co-evaporation [3], co-sputtering [4], electrochemical deposition [5], metal-organic chemical vapour deposition [6] or flash evaporation [7] are some examples. George and Pradeep [8] and Charles et al. [9] reported the deposition of stoichiometric n-type  $Bi_2Te_3$  thin films at flow ratios of Te/Bi > 2 with substrate temperatures between 250 °C and 310 °C. However, films co-evaporated by both groups were n-type only due to an excess Tellurium present in the films. Zou [3] fabricated stoichiometric n-type Bi<sub>2</sub>Te<sub>3</sub> and p-type Sb<sub>2</sub>Te<sub>3</sub> films, and reported some influence of substrate temperature and evaporation rates of the compounds during the film growth process. This work was based on a small number of samples, and no consistent relation could be established between growing conditions and material performance. da Silva and Kaviany [2] reported the fabrication of a microcooler based on co-evaporated films, but due to limitations on fabrication process, optimal growing conditions could not be achieved. To the best knowledge of the authors, no consistent data is available and no report on repeatability of properties of co-evaporated materials performance has been reported to date.



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Fig. 1. Co-evaporation system

In the present work the influence of growing parameters is studied in detail and more than 100 samples were fabricated to allow a consistent correlation of the growing conditions and the thermoelectric properties of the films.

Glass, silicon and polyimide (kapton) were used as substrates, with good film-to-substrate adhesion. However, for thermoelectric applications, 25 µm-thick kapton film was chosen as substrate because of the low thermal conductivity  $(0.12 \text{ W} \text{ m}^{-1} \text{ K}^{-1})$  and appropriate value of thermal expansion coefficient  $(12 \times 10^{-6} \text{ K}^{-1})$ which closely matches the thermal expansion coefficient of the telluride films, thus reducing residual stress and increasing adhesion. Flexible substrates add uncommon mechanical properties to the composite film-substrate and enable their integration with many novel types of devices [10].

# 2. Experimental

Bi<sub>2</sub>Te<sub>3</sub> films were fabricated using the co-evaporation resistive method in a high-vacuum chamber with base pressure <  $5 \times 10^{-6}$  Torr, using two Molybdenum boats, filled with Bismuth pellets and Tellurium lumps, respectively, both with 99.999% purity. The power applied to each boat was controlled independently, using two computed PID controllers to maintain each deposition rate at a fixed value, during the deposition. Bi evaporation rate was maintained at 1 Å/s in all depositions and Te



Fig. 2. XRD analysis of an n-type Bi<sub>2</sub>Te<sub>3</sub> thin film. The peaks agree with the powder diffraction spectrum for Bi2Te3 (dashed lines).

evaporation rate was varied in the range 1-5 Å/s. Each PID controller read the deposition rate from a thickness monitor and was designed to real-time compute the power necessary to apply to the corresponding evaporation boat in order to achieve the userdefined constant evaporation rate. Each thickness monitor, consisting of a quartz crystal oscillator, was carefully placed inside the chamber in order to receive material only from the boat it was monitoring. A metal sheet was placed between the two boats to partially separate the flows from the two evaporants and fully prevent mixing of both materials to occur at the quartz crystals (see Fig. 1). Large boats (baffled boxes, 4 cm<sup>3</sup> volume) were used, in order to achieve stable evaporation rates. A similar procedure was used to evaporate p-type Sb<sub>2</sub>Te<sub>3</sub> thin films by replacing Bi with Sb in one of the boats. Substrates were heated to  $T_{sub} = 240 \degree C$ , 270 °C or 300 °C for Bi<sub>2</sub>Te<sub>3</sub> deposition and to  $T_{sub} = 200 \degree C$  or 240  $\degree C$  for Sb<sub>2</sub>Te<sub>3</sub> deposition. A resistive heater was attached to the substrate holder and temperature was measured on the back of the substrate using a thermocouple embedded in the stainless steel substrate holder. Temperature during deposition was kept constant using a PID controller. A small temperature difference can occur, between the value measured on film surface and the temperature read on the back of the substrate due to low thermal conductivity of polyimide substrate. The film chemical composition and structure were obtained by Energy-Dispersive X-ray spectroscopy (EDX) and Xray diffraction (XRD). In-plane electrical resistivity was measured at room temperature using the conventional four-probe van der Pauw geometry. Seebeck coefficient was measured by connecting one side of the film to a heated metal block at a fixed temperature and the other side to a heat sink at room temperature. Film thickness was around 1 µm for all depositions.

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Properties of selected Bi2Te	3 and Sb <sub>2</sub> Te <sub>3</sub> films	deposited by co-evaporation

Film	Material	Seebeck (µV/°C)	$ ho (\mu\Omega m)$	$T_{\rm sub}$ (°C)	$R = F_{\rm Te}/F_{\rm Bi,Sb}$	Ratio of Te to Bi, Sb (at%)	Composition Te (at%)	P.F. $(\times 10^{-3} \mathrm{W}\mathrm{K}^{-2}\mathrm{m}^{-1})$
#C20	Bi <sub>2</sub> Te <sub>3</sub>	-153	23.0	240	1.55	1.89	65.4	1.0
#C28	Bi <sub>2</sub> Te <sub>3</sub>	-189	19.5	270	1.58	1.51	60.2	1.8
#C30	Bi <sub>2</sub> Te <sub>3</sub>	-111	23.9	270	3.57	2.25	69.2	0.5
#C39	Bi <sub>2</sub> Te <sub>3</sub>	-141	29.5	300	1.57	1.41	58.5	0.7
#C41	Bi <sub>2</sub> Te <sub>3</sub>	-195	16.2	300	2.07	1.50	60.0	2.4
#C59	Sb <sub>2</sub> Te <sub>3</sub>	111	25.7	200	1.67	1.51	60.1	0.5
#C60	Sb <sub>2</sub> Te <sub>3</sub>	124	20.8	200	1.86	1.61	61.7	0.7
#C45	Sb <sub>2</sub> Te <sub>3</sub>	137	24.3	240	1.58	1.74	63.5	0.8
#C44	Sb <sub>2</sub> Te <sub>3</sub>	153	31.5	240	2.72	2.50	71.4	0.7



**Fig. 3.** Seebeck coefficient of  $Bi_2Te_3$  thin films as a function of Te/Bi evaporation rate ratio, *R*, at  $T_{sub} = 240$  °C, 270 °C and 300 °C.

#### 3. Results and discussion

Table 1 summarizes the deposition conditions and measured properties of selected Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> thin films. The evaporation flow ratio, *R*, is defined as the evaporation rate of Te divided by the evaporation rate of Bi or Sb ( $R = F_{Te}/F_{Bi, Sb}$ ). The results show that Te-atom% content is in the range 55–70%, with some films close to stoichiometry (60% Te and 40% Bi/Sb).

#### 3.1. Bi<sub>2</sub>Te<sub>3</sub>

In order to find the deposition parameters which result in films with best thermoelectric properties, three series of Bi<sub>2</sub>Te<sub>3</sub> films were deposited. Bi evaporation rate was 1 Å/s in all evaporations and Te evaporation rate was varied in the range 1–4 Å/s. Fig. 2 shows a typical X-ray diffraction spectrum of an optimized film that reveals its polycrystalline structure. The peaks agree with the powder diffraction spectra for polycrystalline Bi<sub>2</sub>Te<sub>3</sub>. Fig. 3 shows the Seebeck coefficient of all Bi<sub>2</sub>Te<sub>3</sub> films, for the three different substrate temperatures used – 240 °C, 270 °C and 300 °C – as a function of the Te/Bi evaporation ratio *R*.

For the same *R*-value films have higher Seebeck coefficient for higher substrate temperature (300 °C) than for lower substrate temperature (270 °C, 240 °C). At  $T_{sub} = 300$  °C the optimal evaporation rate of Te is 2.0 times higher than the evaporation rate of Bi, while at  $T_{sub} = 240$  °C, *R* is only 1.6. This could be attributed to reevaporation of Te from the substrate as discussed in next paragraph.



**Fig. 4.** Ratio of Te/Bi number of atoms in the films as a function of Te/Bi evaporation rate ratio, *R*, for  $T_{sub} = 240$  °C, 270 °C and 300 °C.



Fig. 5. XRD analysis of a p-type  $Sb_2Te_3$  thin film. The peaks agree with the powder diffraction spectrum for  $Sb_2Te_3$  (dashed lines).

In Fig. 4, the value of the Bi<sub>2</sub>Te<sub>3</sub> film composition (number of Te atoms divided by number of Bi atoms) obtained by EDX is plotted against *R* for different substrate temperatures (240 °C, 270 °C and 300 °C). It can be seen that film composition is always lower in Te than the corresponding fraction of evaporated Te, *R*. This effect is more important as  $T_{sub}$  increases from 240 °C to 270 °C and to 300 °C. At 300 °C, the vapour pressure of Te, taken from literature, is 10<sup>5</sup> higher than the vapour pressure of Bi. Therefore, it is possible to conclude that re-evaporation of Te from the substrate at a higher rate than Bi explains why it is necessary to use a higher value of *R* than the desired ratio of atoms of Te/Bi in the film final composition. In order to quantify this effect several evaporations of pure Te were made at 300 °C on polyimide and the conclusion is that, at this  $T_{sub}$ , Te re-evaporation rate can be as-high-as 2 Å/s.

If one looks at the optimized values of Seebeck coefficient as a function of film composition it can be seen (Table 1) that the best films are those whose composition is close to stoichiometry.

## 3.2. Sb<sub>2</sub>Te<sub>3</sub>

A similar procedure to the one used to optimize Bi<sub>2</sub>Te<sub>3</sub> films was adopted for Sb<sub>2</sub>Te<sub>3</sub>. Two deposition series were made at  $T_{sub} = 200$  °C



Fig. 6. Seebeck coefficient of  $Sb_2Te_3$  films fabricated at different Te/Sb evaporation fluxes ratio and different substrate temperatures.



**Fig. 7.** Composition ratio of Te/Sb number of atoms in the film as a function of *R*, for  $T_{sub} = 200 \text{ }^{\circ}\text{C}$  and 240  $^{\circ}\text{C}$ .

and 240 °C. In both T<sub>sub</sub>-series Sb evaporation rate was fixed at 1 Å/s and Te evaporation rate was varied in the range 1-4 Å/s. Fig. 5 shows a typical X-ray diffraction spectrum of an optimized film that reveals its polycrystalline structure. The peaks agree with the powder diffraction spectra for polycrystalline Sb<sub>2</sub>Te<sub>3</sub>. Fig. 6 shows the Seebeck coefficient measured in films deposited at different R = Te/Sb evaporation ratios, for  $T_{sub} = 200 \,^{\circ}\text{C}$  and 240  $^{\circ}\text{C}$ . The higher Seebeck coefficients are obtained for R = 2.0 - 2.5. However, in this case and contrary to what was observed for Bi<sub>2</sub>Te<sub>3</sub>, temperature of substrate in the range studied has little or no influence on Seebeck coefficient. Also, the re-evaporation effect of Te from the substrate is much less important or is totally absent here, as can be seen in Fig. 7 where film composition roughly reproduces the evaporation ratio R used for deposition. This could also be related to the lower  $T_{sub}$  used for Sb<sub>2</sub>Te<sub>3</sub> than for Bi<sub>2</sub>Te<sub>3</sub> deposition. Sb<sub>2</sub>Te<sub>3</sub> films with higher Seebeck coefficient are slightly rich in Te (no. atoms Te/no. atoms Sb = 2-2.5).

# 4. Conclusions

State-of-the-art Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> materials were deposited by co-evaporation, in the form of thin films. EDX results show that Bi<sub>2</sub>Te<sub>3</sub> films with high Seebeck coefficients are obtained when the composition of film is stoichiometric (60% Te and 40% Bi). With lower substrate temperatures (240 °C), best films are obtained with Te/Bi evaporation flow ratio of 1.6, but at higher substrate temperatures

 $(300 \ ^{\circ}C)$  the evaporation flow ratio should be increased to 2.0, due to re-evaporation of Te from the heated substrate. The best Sb<sub>2</sub>Te<sub>3</sub> films are obtained with Te/Sb evaporation flow ratio in the range 2.0–2.5. Substrate temperature, in the range 200–240  $^{\circ}C$ , has no major influence on Sb<sub>2</sub>Te<sub>3</sub> Seebeck coefficient.

Since small deviations on evaporation rates result in large differences in material properties, an accurate calibration of thickness monitor parameters should be done, adjusting the density, Z-factor and tooling factor. Use of as-high-as possible temperature of substrate is also recommended, since the excess of Tellurium will be easily re-evaporated. Substrate temperature above 300 °C becomes impracticable, due to excessive re-evaporation of Tellurium from substrate.

Compatibility with flexible electronics was demonstrated by the use of a 25  $\mu$ m-thick polyimide foil as substrate, due to its low thermal conductivity and high upper working temperature. Thermoelectric properties achieved on Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> thin films deposited on polyimide in this work are suitable for the fabrication of Peltier microcoolers and thermoelectric microgenerators.

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